The influence of phosphate on Fe(II) catalyzed ferrihydrite transformation under oscillating redox condition

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## Background

Ferric (hydr)oxides (FHO) play a key role in controlling redox active species

Dynamic environments experience repeated redox fluctuations

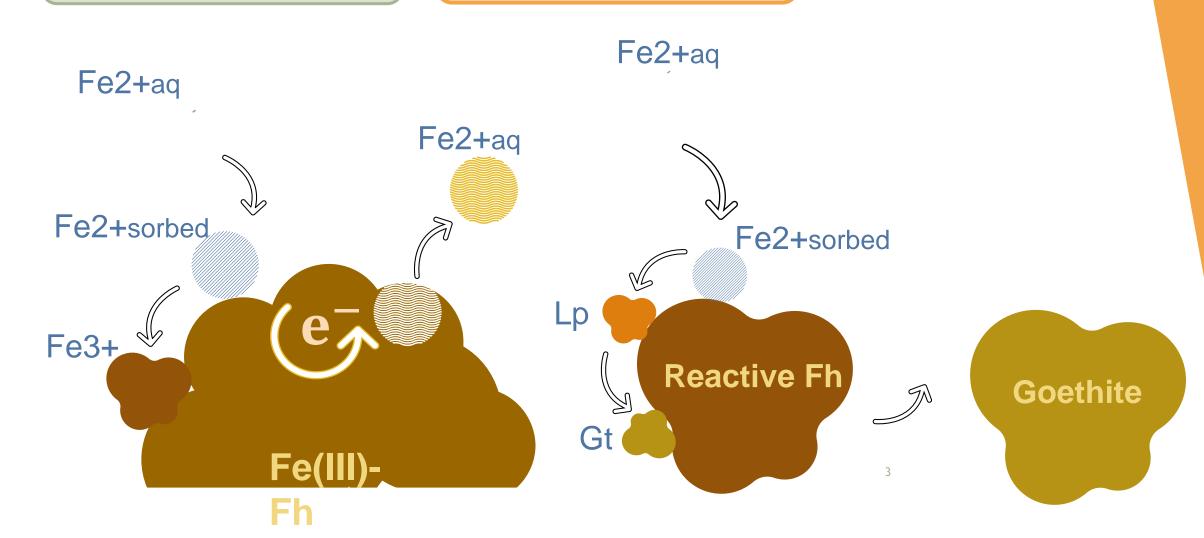
prominent effect of Fe on the mobility of nutrients and contaminants at hydrological interfaces



### Fe2+ catalyzed Fh transformation

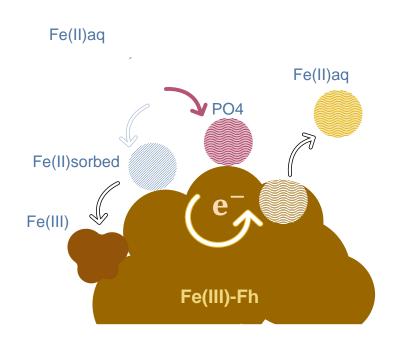
Transformation of Fh to more crystalline phase

electron transfer



### Impact of PO4 and redox oscillation

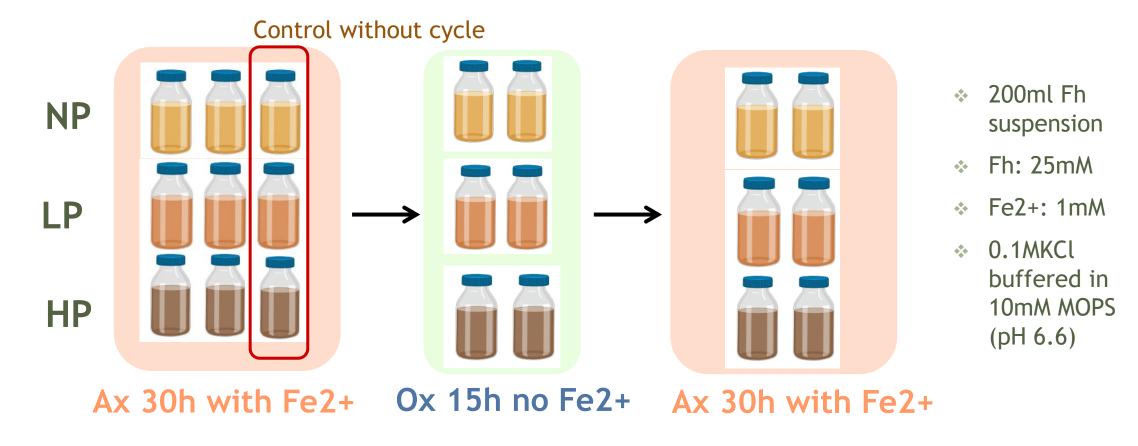
- Impact of PO4 adsorption
  - ▶ Passivation of Fh surface
  - Decrease Fh dissoltuion rate
  - On Fe2+ catalyzed Fh transformation?



#### **Objective:**

- -investigate the influence of phosphate on Fe2+ catalyzed ferrihydrite transformation.
- -determine the impact of fluctuating redox condition over 5 short redox cycles at two P concentrations

#### Experimental set up



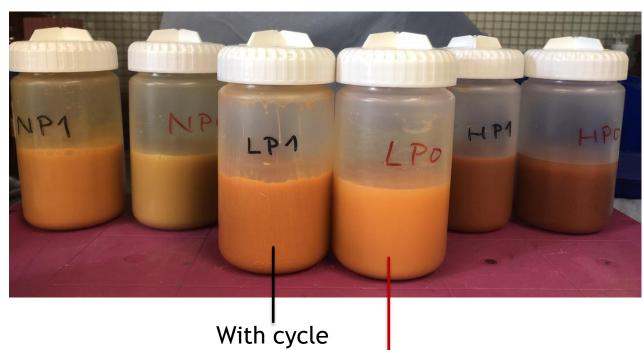
Synthesized Fh by base titration

LP: 0.05mM, Fe/P= 500

 $\blacksquare$  HP: 0.5 mM P, Fe/P = 50







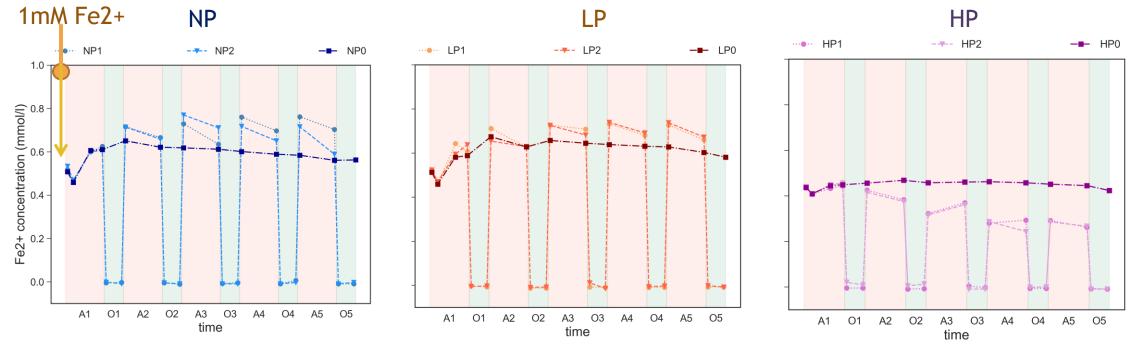
control

# Color development

- Top left: start of Exp
- Top right: start of 2<sup>nd</sup> cycle

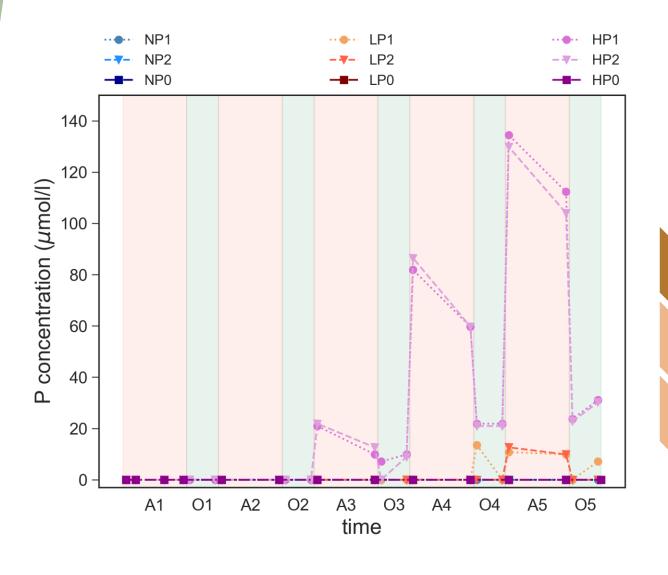
Down: end of Exp

#### Temporal trends in aq. Fe2+



- Initial Fe2+ uptake by Fh (spiked Fe2+: 1mM)
- Rapid change of Fe2+ during first hours before reaching equilibrium
- ► Higher Fe2+ uptake in controls (NP, LP)
- ► HP: enhanced Fe2+ removal by redox cycles
  - ► Adsorption or precipitation?

## Temporal trends in aq. P



► HP: 0.5 mM P

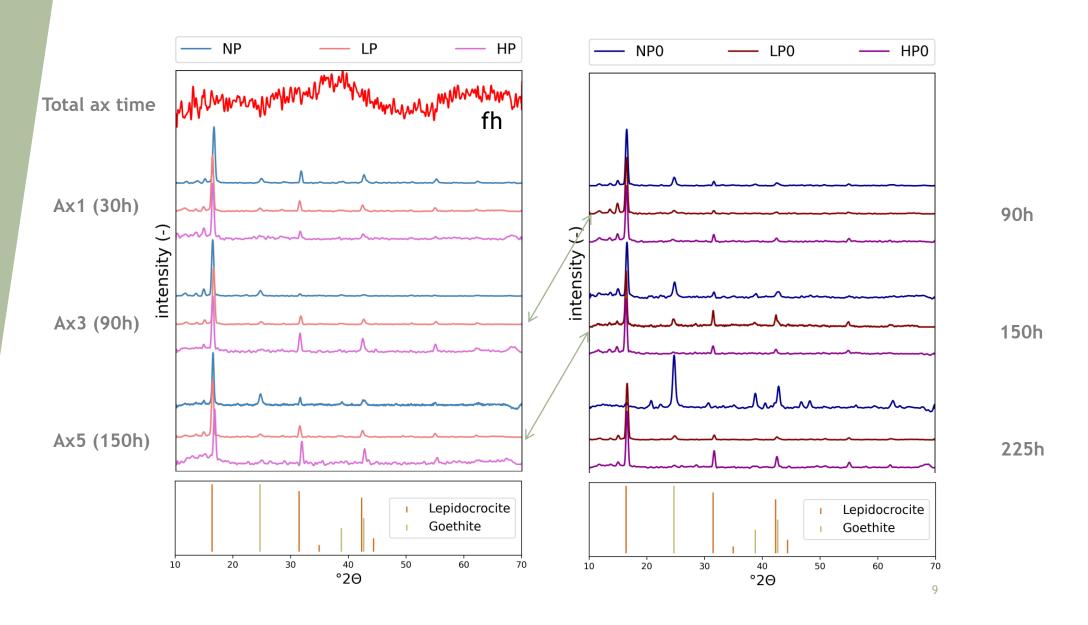
► LP: 0.05 mM P

• Confirm Mineral transformation

• Alter P and Fe2+ sorption

• Influence electron transfer

#### Evolution of mineral phases by Powder X-ray diffraction (XRD)



### Summary



- Redox fluctuations largely impede the transformation
- ✓ Better understand biogeochemical cycling of Fe and P in redox active environments
- Filter material for P-removal under anoxic condition

# Thank you